

Synthesis of 12-Alkyl Analogues of Prostaglandin-A₂

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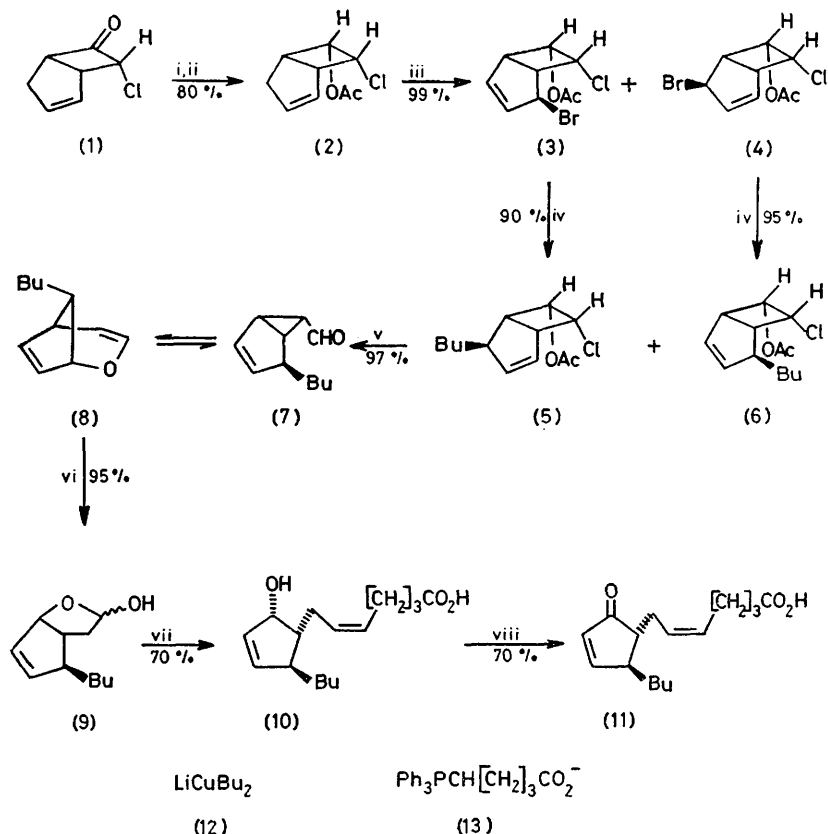
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Summary The prostaglandin-A₂ analogue (**11**) has been prepared *via* an *SN'* *syn* reaction involving the halogenoesters (**3**) and (**4**) with a butyl cuprate reagent.

In view of the biological interest in prostaglandin A congeners,¹ we required a synthetic route to prostaglandin A₂

analogues in which the octenol side chain was replaced by a simple alkyl group. Our new method of preparation of these compounds is illustrated by the synthesis of the butyl-compound (**11**) which involves eight steps from the readily available chloroketone (**1**)² and an overall yield of 35%.



Reagents: i, NaBH_4 , EtOH. ii, Ac_2O , $\text{C}_6\text{H}_5\text{N}$. iii, *N*-Bromo-succinimide, *hv*, CCl_4 . iv, Reagent (12), -78°C , tetrahydro-furan. v, NaOMe-MeOH . vi, $(\text{CO}_2\text{H})_2\text{-H}_2\text{O}$. vii, Reagent (13). viii, Collins oxidation.

Regiospecific borohydride reduction of the ketone (1)^{3,4} followed by acetylation furnished the ester (2) which on photon-induced bromination employing *N*-bromosuccinimide (NBS) in carbon tetrachloride gave a mixture of the dihalogenoacetates (3) and (4). This mixture reacted with the cuprate reagent (12) in tetrahydrofuran at -78 °C to give a mixture of the butylbicycloheptenones (5) and (6).⁵ Sodium methoxide in methanol at room temperature converted this mixture into the cyclopropylaldehyde (7) which exists in equilibrium with the 2-oxabicyclo[3.2.1]-octa-3,6-diene (8) at ambient temperature.⁶ An aqueous solution of oxalic acid effected hydrolysis of the cyclic enol

(8) and furnished the lactol (9) in practically quantitative yields.⁷ A Wittig reaction involving the lactol (9) and the reagent (13) gave the prostanoid (10) which was subjected to a Jones oxidation procedure to form the desired cyclopentenone derivative (11).

We observed that the compound (11) possesses potent analgaesic properties.

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¹ J. R. Weeks, *Science*, 1973, **181**, 370.

² For an alternative synthetic route to related prostaglandin-A₁ analogues see J. B. Wiel and F. Rouessac, *J.C.S. Chem. Comm.*, 1976, 446.

³ M. Rey, S. M. Roberts, A. Dieffenbacher, and A. S. Dreiding, *Helv. Chim. Acta*, 1970, **53**, 417.

⁴ P. R. Brook, A. J. Duke, and J. R. C. Duke, *J.C.S. Chem. Comm.*, 1970, 574.

⁵ C. B. Chapleo, M. A. W. Finch, T. V. Lee, S. M. Roberts, and R. F. Newton, preceding communication.

⁶ *cf.* M. Rey and A. S. Dreiding, *Helv. Chim. Acta*, 1965, **48**, 1987; P. R. Brook, A. J. Duke, J. M. Harrison, M. Rey, S. M. Roberts, and A. S. Dreiding, *Helv. Chim. Acta*, 1977, **60**, 1528; see also K. Harding and J. M. Trotter, *J. Org. Chem.*, 1977, **42**, 4157.

⁷ These reaction conditions were suggested by M. Rey and A. S. Dreiding from their unpublished work on related systems.